### Aquatic Chemistry for Engineers

# Modeling water treatment trains and processes with PHREEQC

2 April 2015

Peter de Moel – Omnisys/TU Delft





### Personal introduction ... Record on water chemistry

- 1979 1980 : KIWA (now KWR)
  - KIWA report and computer program aggressive water
- 1980 2000 : DHV (now RHDHV)
  - Patent softening Amsterdam (Graveland cs)
  - Publication / HP41 program on CaCO3 equilibrium
  - Publications / presentations (H2O JAWWA Las Vegas)
  - Design and build over 40 water treatment plants, wordwide
- 2000 present : TU Delft (part-time)
  - BSc / MSc education
  - OpenCourseWare (2007 2014)
  - Aquatic Chemistry 4 Eng (2011 now)
  - Online MSc (2012 2014)
  - edX MOOC (2013 2014)

**ŤU**Delft



CTB3365x: Introduction to Water Treatment Learn about urban water services, focusing on basic drinking water and wastewater treatment technologies MORE STARTS: 16 Sep 2013 • INSTRUCTORS: Jules van Lier • DelftX





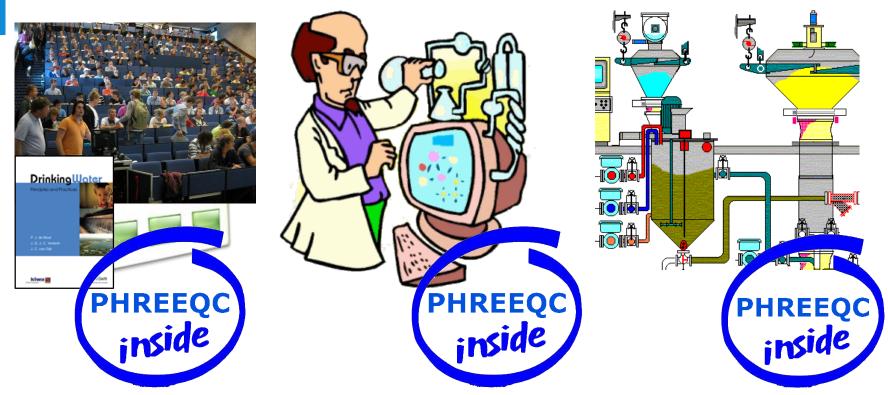


### AC4E – Our focus points

**Education** 

Research

**Process control** 

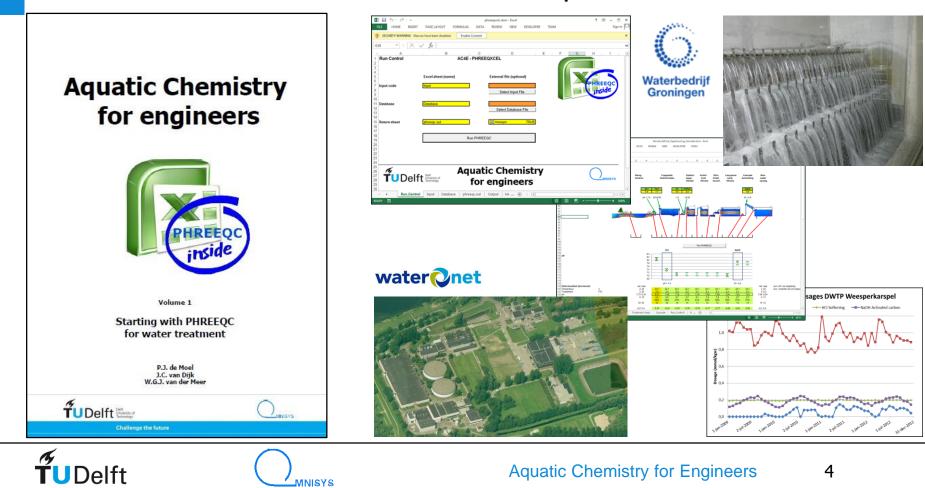


Practical applications for drinking water, waste water and industrial water





# PhreeqXcel – PHREEQC in Excel with Stimela.dat



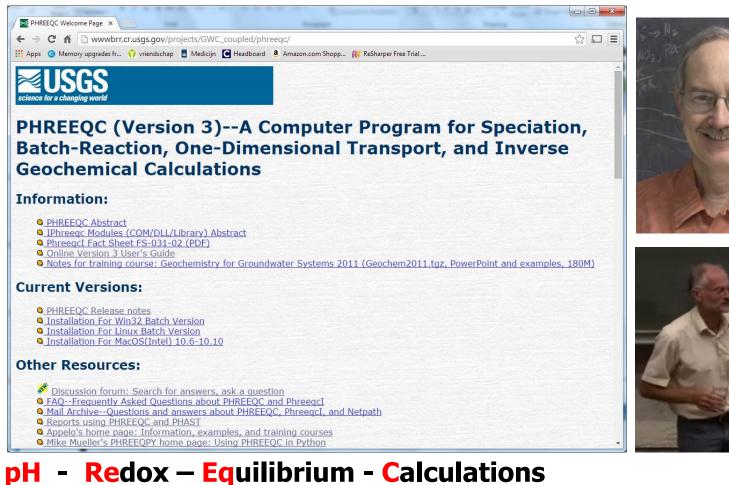
2<sup>nd</sup> IWA New Developments in IT & Water Conference

### PHREEQC

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#### Development since 1980

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### PHREEQC

pH - Redox - Equilibrium Calculations

- 25 elements
- 21 redox states
- 180 species
- 72 solid phases
- 8 gas phases
- + Exchange eq.
- + Surface eq.
- + Rates

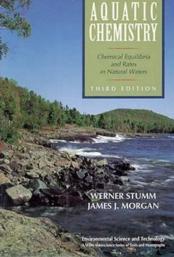
**TU**Delft

8 databases (+ Stimela.dat)

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#### Aquatic Chemistry (Stumm & Morgan) in your laptop

C Ca Mg Na etc C(+4) C(-4) etc C(+4) : CO2 HCO3 CO3 etc C(-4) : CH4 etc CaCO3 Fe(OH)3 etc CO2 H2S etc (ion-exchange) (activated carbon) (kinetics)





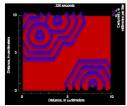
### PHREEQC Freely available

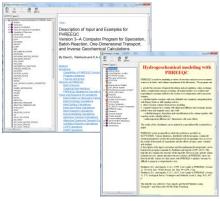
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Download:					Batch-Reactio	
				ar files available below. However, some World Wide Web ated in the installation instructions may be unnecessary.	Chapter 43 of Section A, Ground Book 6, Modeling	
		Graphical Use	r Inte	faces	10 _ 1	
Platform	Processor	File names	Size	Notes		
		phreeqci-3.1.7-9213.msi	13M	Executable, source, database files, examples, PDF documentat	tion	
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		PHREEQC for Windows		PHREEQC for Windows Home Page		
		Batch Ve	rsions	к		
Platform	Processor	File names	Size	Notes		
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Linux	64-bit	phreeqc-3.1.7-9213.x86 64.tar.gz	20M	Executable, database files, examples, PDF documentation		
	Source	phreeqc-3.1.7-9213.tar.gz	12M	Source, database files, examples, PDF documentation	U.S. Department of the Int U.S. Geological Survey	lerior
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xamples for PHREEQC ogram for Speciation, nsional Transport, and ulations





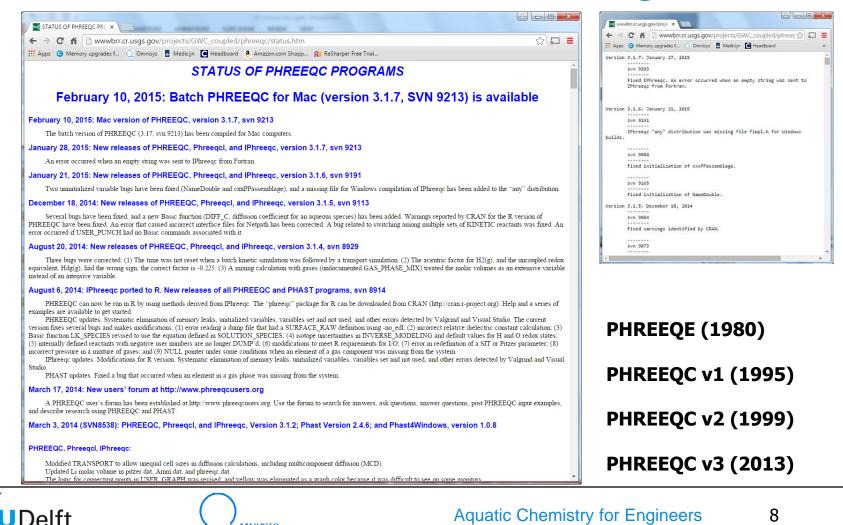
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### PHREEQC

#### Latest release PHREEQC: 1.5 months ago

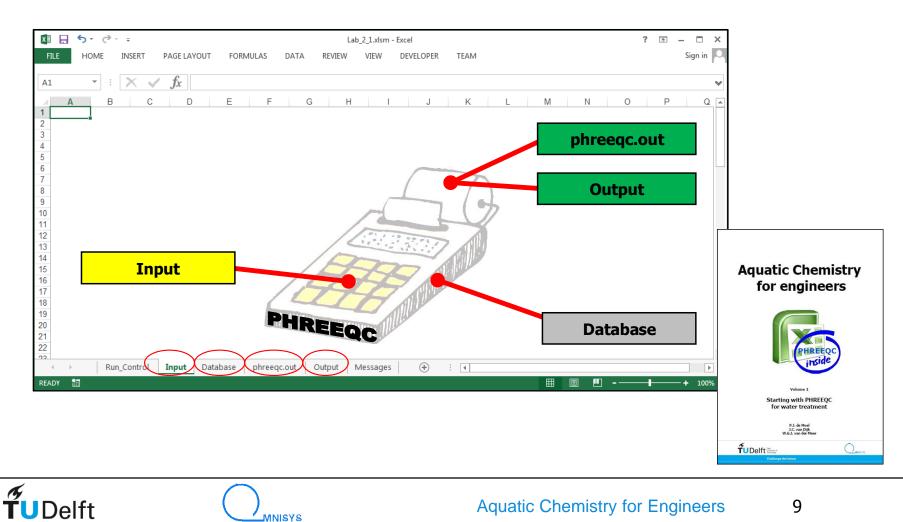
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### PHREEQXCEL

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Excel with IphreeqcCOM for Windows + Stimela.dat

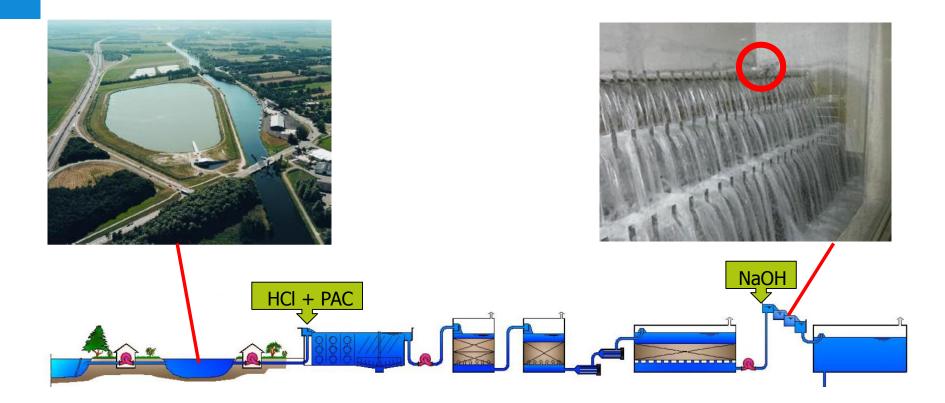


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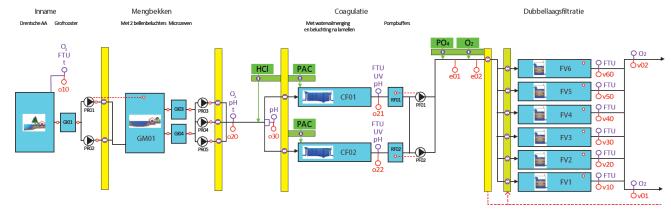


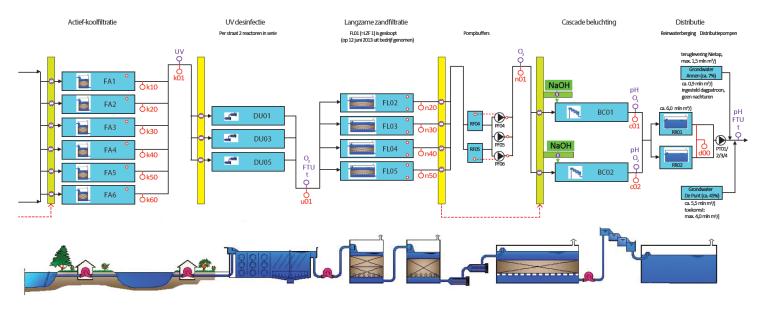


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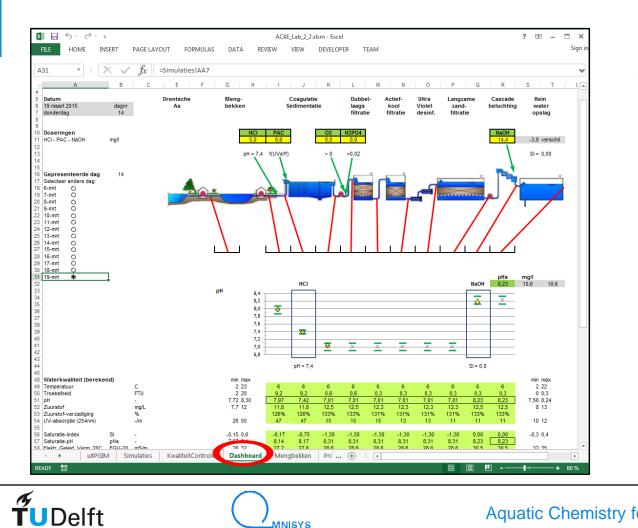




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#### **Modeling for** 'day-average' values

10 process/sample points

31 water quality parameters

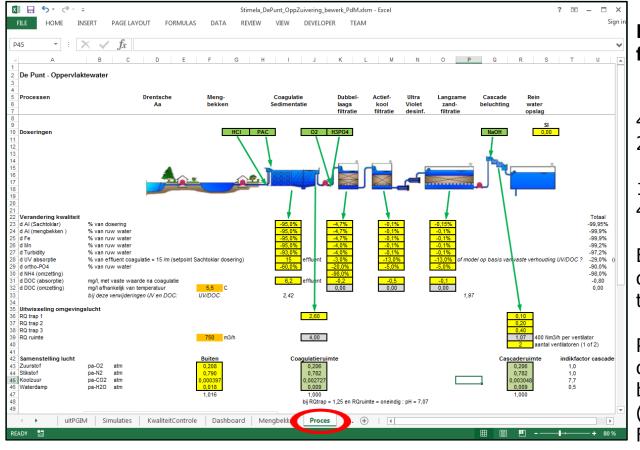
14 days

=

4,340 numbers / batch

Selected per day for presentation





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#### Removal efficiencies from historical data

4 removal processes 2 gas exchange processes

10 water quality parameters 4 gas/air quality parameters

Biological TOC conversion depends on water temperature

Parameter to be determined/modified by process engineer (yearly ?) Future: self learning system





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61	water (Influent) DN_SPREAD				
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65 66 # 67 #	-water Oxg Ntg	1 #kg weter 10xg(g) -0.6819 #\$ling(g)allog(0,208) -0.6819 18kg(g) -0.0024 #\$ling(g)allog(0,78) -0.024			
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75 76 TITLE	Calculation pe :	nd pH at field temperature			·
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82 83 TITLE	Dosing HCI				
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STITLE	Coagulation USE solution Coagulation	Dosieg and removal Sacktokhar Addia; AUXI33(SO4)0.145(OH)1.38 Penovak (x dozina) 2 ; REACTION1 ; AUXI33(SO4)0.145(OH)138 IX; AUXI35(SO4)0.15(OH)138 (AUXI35)000 ; 10,244444 mmoles Removal zwa water composate Ali (X Rwij Fgl/z Rwij Mal/z Rowij Doc (x Rwij etc.	; SAVE solution 3 ; END	# Simulation	5
9 0 TITLE	USE solution Coagulation	3 ;FEACTION1 ;AI(ON) - 0.00558 ;F(ON)3-0.00558; MMO2 -0.00128; Turb :5551400;Doc -0.45201;Uvab ;5151000;PO4 -0.00158; 10 mmoles	; SAVE solution 3 ; END	# Simulation	6
1 TITLE	USE solution Coagulation	3 ;GAS_PHASE 1, fixed_pressure ; pressure 100 ; "volume 400000"; ; temperature 55::10; ;C2(g) 1028000(; Rkg(g) 11780000; ;C02(g) 10.000035; ;H20(g) 11.008000 Acration at certiflow Acration increase organge, decrease acrosso discide, increase organge, decrease acrosso discide, increase organge, decrease acrosso discide, increase organized acrosso discide acrosso	; END	# Simulation	7
93 94 95 TITLE	USE solution Dosing 02 + H3	3 ; GAS_PHASE t; fixed_pressure ; pressure 100 ; rolome 20000; ; temperature 55 ; ; ; ; ; ; (2(g) 0,20002; ; Wg(g) 0,12209; ; ; CO2(g) 0,002122; ;H2O(g) 0,003152	; SAVE solution 3 ; END	# Simulation	8
16 17	USE solution	3 ;REACTION1 ; O2 <mark>0,000000;</mark> ;KSPO4 0,000000 ;1.0 mmoles	; SAVE solution 4 ; END	# Simulation	9
98 TITLE 99	USE solution	Removal Al-Sacktoblar and raw water compounds Al (xX RawyX Sackt.) Fe(xX Raw) Mn(xX Raw) Doc (xX Raw) etc 4 ;REACTION 1 ;Al(OH)3 <mark>-0.01174</mark> ;Fe(OH)3 <mark>-0.01058</mark> ;MnO2 <mark>-0.00005</mark> ;Twb <mark>:0.06612</mark> 0 ;Doc <mark>:0.016551</mark> ;Uvab <mark>:0.45000</mark> ;PO4 <mark>-0.00051</mark> ;1.0 mmoles	; SAVE solution 5 ; END	# Simulation	10
0 TITLE	USE solution	Conversion NH4 Convertinert NH4 in redox-NH4 = N(-3) for starting oxyudition reactions 5 ; REACTION 1 ; [N-3]H4 -L0; ; NH4 = 10; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ; ;	; SAVE solution 5 ; END	# Simulation	11
2 TITLE 3 4	USE solution	Cosrersion DOC Convertinent Doc with 02 into CO2 5 ;REACTION 1 ; Doc -1.0; O2 -1.0; CO2 1.0; 10,000000 mmoles	; SAVE solution 5 ; END	# Simulation	12
5 TITLE	AC filtration USE solution	Removal Al-Sacktoblar and raw water compounds         Al (x2 Raw y0, Sacht.)         Fe(x2 Raw)         Mn(x2 Raw)         Dec (x2 Raw)         Dec	; SAVE solution 6 ; END	# Simulation	13
07 TITLE	AC filtration USE solution	Conversion DOC Convert inert Doc with 02 into C02 6 ;REACTION 1; Doc -1.0; O2 -1.0; C02 1.0; (0,000000 mmoles	; SAVE solution 6 ; END	# Simulation	14
03 0 #	UV				
12 TITLE	SS filtration USE solution	Removal Al-Sacktoklar and raw water composeds Al (x8 Rownys Sackt) Fe(x8 Row) Mn(x8 Row) Occ (x8 Row) Atc 6 (x8 Row) Atc 6 (x8 Row) Atc 7 (x8	; SAVE solution 7 ; END	# Simulation	15
4 TITLE	SS filtration USE solution	Conversion DOC Convert inst Doc with 02 into CO2 7 ;REACTION 1; Doc -1.0; O2 -1.0; CO2 1.0; <b>10,00000</b> mmoles	; SAVE solution 7 ; END	# Simulation	16
6 7 TITLE 8	Acration USE solution	Composition air in cascade compartments 7 : GA3_PHASE : +Trod_pressure :-pressure 1.00 :+volume <b>10811 ::</b> :+temperature <b>85:::::</b> :02(g) <mark>102080000</mark> :Ntg(g) <mark>10,0000000</mark> :CO2(g) <b>10,0000000</b> :H2O(g) <b>10,000000</b>	: END	# Simulation	17
13 TITLE 20	Acration USE solution	i UAS_PHASE 1 Price_pressive : Decisive : pressive : DUI == volume : Lobert : : : chemperature & st.::::: [Udig] : UZARDOMI : Volg] : UZARDOMI : : UDIg] : UZARDOMI : : : UDIg] : UZARDOMI : : : UDIg] : UZARDOMI : : : : : : : : : : : : : : : : : : :	; END ; SAVE solution 8 ; END	# Simulation	18
21 TITLE	Acration USE solution	Aerations in setep 2 Aerations: increase oxygen, decrease curbon dioxide, increase/decrease/introgen 6 _ 10.8PMAS1: f:/meteoreser: _pre-reserve 10.0 ; rouber = 0.2000; ::enaporature :	;SAVE solution 8 ;END	# Simulation	19
23 TITLE 24	Acration USE solution	Aeration in step 3 Aeration: increase oxygen, decrease curbon dioxide, increaseldecrease nitrogen 8 ;GAS_PHASE 1;-fixed_pressure ; pressure 100;+rolume 104000;;; remperature 85;;;; C2(g) 11205775; Wrg(g) 10786989; CO2(g) 11.003044; H2O(g) 10.003044;	; SAVE solution 8 ; END	# Simulation	20
25 26 TITLE 27	Dosing NaOH USE solution	Required dose for \$1=0 8 ;EQUILIBRIUM_PHASES 2 ;Caloite 10,0000 <sup>2</sup> , NoOH 0.01	; SAVE solution 3 ; END	# Simulation	21
28 TITLE 29	Calculation USE solution USE solution		; SAVE solution 3 ; END ;	# Simulation	21
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#### **PHREEQC** input code

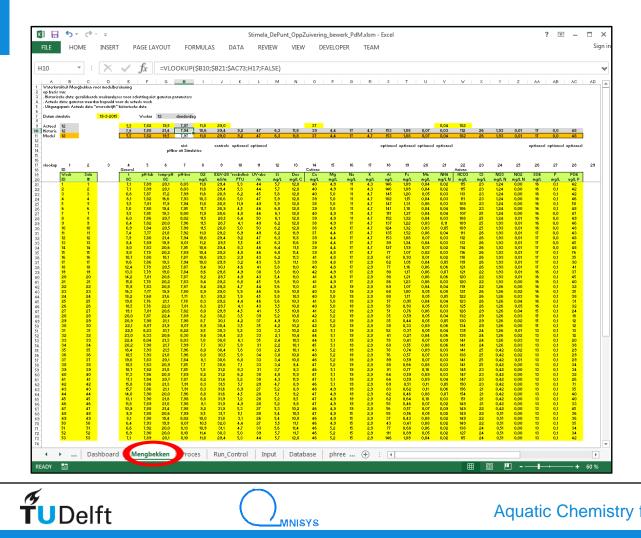
Simulation per day

Orange cells: Raw water quality from reservoir (from sheet Mengbekken)

Yellow cells: Process parameters (from sheet Proces)

1-3 program code linesper treatment process(22 PHREEQC calculations)





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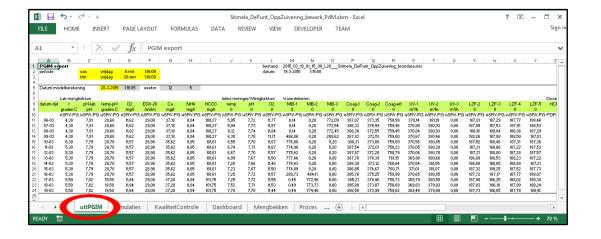
#### Actual and historical water quality from reservoir

9 actual quality parameters (weekly sampling) 25 historical quality parameters from previous year(s)

Actual overrules historical Actual compared with historical

Validation checks





#### Daily export from process computer (automated)

14 days `day-average' values

Each day:

8 water quality parameters

(measured by lab)

27 online water quality

parameters/location

9 online flow measurements (water)

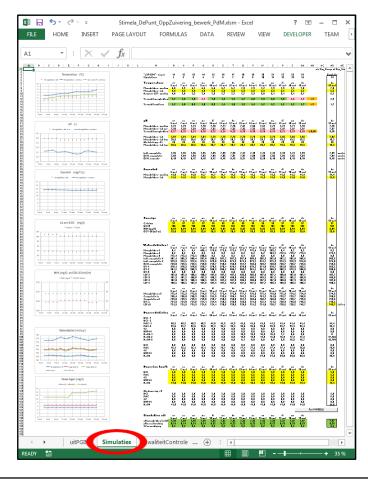
13 online flow measurements (chemicals)

672 numbers / batch









#### Simulations per day

14 days 'day-average' values

Model input for: Water flows Chemical flows

Calculated: Dosing levels pH at actual temperature O2-saturation (%) pHs (setpoint NaOH)

Graphical presentation







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#### **Quality control for** operators

14 days overview

'Green zones' for acceptable values

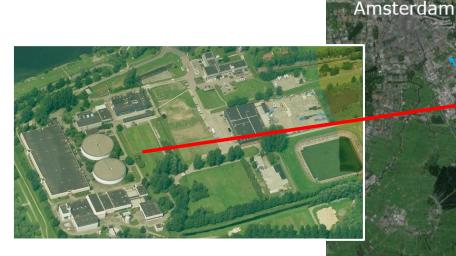
Accessible for modifications by operators, supervisors, process engineers, etc...

- no specific skills
- no proprietary equipment

Graphical presentation

### waterQnet

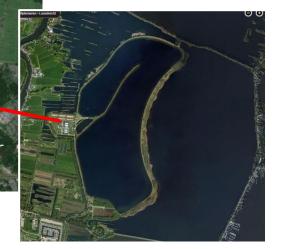
### Drinking water - Weesperkarspel



Drinking water treatment plant Weesperkarspel

> Pre-treatment Loenderveen

> > Source Bethunepolder







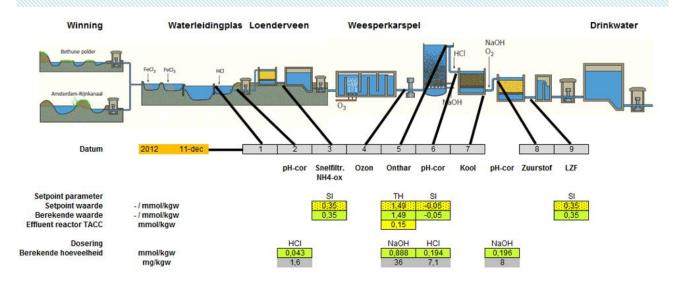


### Drinking water - Weesperkarspel





### SI and Total Hardness (TH)



4 influencing processes in the treatment





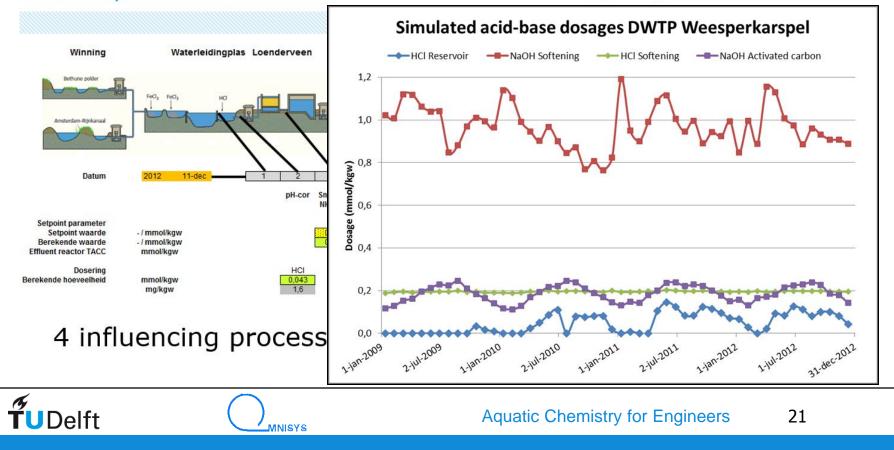
### waterQnet

### Drinking water - Weesperkarspel

/innovation



### SI and Total Hardness (TH)





### Aquatic Chemistry for Engineers Further information

#### **OpenCourseWare website**

- <u>http://drinkwater.citg.tudelft.nl/AquaticChemistry</u>
- http://www.omnisys.nl

#### Contains

- Lectures
- Readings
- Activities (Labs and Tests)
- New developments
- Database stimela.dat (phreeqc.dat for Water Treatment, updated)











### PHREEQC

#### Drinking water in PHREEQC – in the cloud

Drinking water - Conductivity + Charge balance
Conductivity (EC) + Charge balance

Steps to do:

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- Fill in water quality data
- Press Run Phreego
- Wait a few seconds for (updated) output
- optional: Change input values and re-Run Phreeqc
- optional: if email-address is filled: Send Excel file (full version) by email optional: if email-address is not filled: Download and Save Excel file (full version)

Temperature	t	°C	11,5
Oxygen	Oz	mg/L	11,0
pH			7,91
Conductivity (EC 20 ° C)		mS/m	38,4
Cations			
Calcium	Ca	mg/L	40,5
Magnesium	Mg	mg/L	5,30
Sodium	Na	mg/L	49,7
Potassium	к	mg/L	2,0
Anions			
Hydrogen carbonate	HCO3	mg/L	199
Chloride	Cl	mg/L	28
Nitrate	NO <sub>3</sub>	mg/L	7,0
Sulfate	SO₄	mg/L	7,9

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Anions m Conductivity (EC at t) m Total dissolved solids (TDS) m Ionic strength m Total hardness m Redox conditions pe (electron activity) Redox potential m Correctness checks Charge difference m Percentage error (100°(Cat- An )/(Cat+  EC ratio, calculated/measured pH change by electron balancing (Phreeq Carbon equilibrium pH (Hydrogen activity) Alkalinity m Total inorganic carbon (TIC) m HCO <sub>2</sub> - m		4,61 4,26 31,4 339 5,7 1,23 13,92 785 0,34 3,85 % 1,01 0,000
Anions m Conductivity (EC at t) m Total dissolved solids (TDS) m Ionic strength m Total hardness m Redox conditions pe (electron activity) Redox potential m Correctness checks Charge difference m Percentage error (100*(Cat- An )/(Cat+  EC ratio, calculated/measured pH change by electron balancing (Phreeq Carbon equilibrium pH (Hydrogen activity) Alkalinity m Total inorganic carbon (TIC) m HCO <sub>2</sub> - m	neq/kgw n5/m ng/L nmol/kgw nmol/kgw nw nv nv nv neq/kgw (An   )	4,26 31,4 339 5,7 1,23 13,92 785 0,34 3,85 % 1,01 0,000
Conductivity (EC at t) m Total dissolved solids (TDS) m Ionic strength m Total hardness m Redox conditions pe (electron activity) Redox potential m Correctness checks Charge difference m Percentage error (100*(Cat- An )/(Cat+  EC ratio, calculated/measured pH change by electron balancing (Phreeq Carbon equilibrium pH (Hydrogen activity) Alkalinity m Total inorganic carbon (TIC) m HCO <sub>2</sub> - m	nS/m ng/L nmol/kgw nwol/kgw nV nv neq/kgw (An  )	31,4 339 5,7 1,23 13,92 785 0,34 3,85 % 1,01 0,000
Total dissolved solids (TDS) m Ionic strength m Total hardness m Redox conditions pe (electron activity) Redox potential m Correctness checks Charge difference m Percentage error (100*(Cat- An )/(Cat+  EC ratio, calculated/measured pH change by electron balancing (Phreeq Carbon equilibrium pH (Hydrogen activity) Alkalinity m Total inorganic carbon (TIC) m HCO <sub>2</sub> - m	ng/L Imol/kgw Imol/kgw IV Inv Ineq/kgw (An   )	339 5,7 1,23 13,92 785 0,34 3,85 % 1,01 0,000
Ionic strength m Total hardness m Redox conditions pe (electron activity) Redox potential m Correctness checks Charge difference m Percentage error (100*(Cat- An )/(Cat+  EC ratio, calculated/measured pH change by electron balancing (Phreeq Carbon equilibrium pH (Hydrogen activity) Alkalinity m Total inorganic carbon (TIC) m HCO <sub>3</sub> - m	imol/kgw imol/kgw iV neq/kgw [An])	5,7 1,23 13,92 785 0,34 3,85 % 1,01 0,000
Total hardness       m         Redox conditions       m         pe (electron activity)       m         Redox potential       m         Correctness checks       m         Charge difference       m         Percentage error (100*(Cat- An )/(Cat+ EC ratio, calculated/measured pH change by electron balancing (Phreeq         Carbon equilibrium         pH (Hydrogen activity)         Alkalinity       m         Total inorganic carbon (TIC)       m         HCO <sub>2</sub> -       m	nmol/kgw nV neq/kgw (An))	1,23 13,92 785 0,34 3,85 % 1,01 0,000
pe (electron activity) Redox potential m Correctness checks Charge difference m Percentage error (100 <sup>*</sup> (Cat- An )/(Cat+  EC ratio, calculated/measured pH change by electron balancing (Phreeq Carbon equilibrium pH (Hydrogen activity) Alkalinity m Total inorganic carbon (TIC) m HCO <sub>2</sub> - m	ieq/kgw  An )	0,34 3,85 % 1,01 0,000
Redox potential     m       Correctness checks     m       Charge difference     m       Percentage error (100*(Cat- An )/(Cat+        EC ratio, calculated/measured       pH change by electron balancing (Phreeq       Carbon equilibrium       pH (Hydrogen activity)       Alkalinity     m       Total inorganic carbon (TIC)     m       HCO <sub>3</sub> -     m	ieq/kgw  An )	0,34 3,85 % 1,01 0,000
Correctness checks Charge difference m Percentage error (100 <sup>*</sup> (Cat- An )/(Cat+  EC ratio, calculated/measured pH change by electron balancing (Phreeq Carbon equilibrium pH (Hydrogen activity) Alkalinity m Total inorganic carbon (TIC) m HCO <sub>3</sub> - m	ieq/kgw  An )	0,34 3,85 % 1,01 0,000
Charge difference m Percentage error (100*(Cat- An )/(Cat+  EC ratio, calculated/measured pH change by electron balancing (Phreeq Carbon equilibrium pH (Hydrogen activity) Alkalinity m Total inorganic carbon (TIC) m HCO <sub>2</sub> - m	An)	3,85 % 1,01 0,000
Percentage error (100*(Cat- An )/(Cat+  EC ratio, calculated/measured pH change by electron balancing (Phreeq Carbon equilibrium pH (Hydrogen activity) Alkalinity m Total inorganic carbon (TIC) m HCO <sub>2</sub> - m	An)	3,85 % 1,01 0,000
EC ratio, calculated/measured pH change by electron balancing (Phreeq Carbon equilibrium pH (Hydrogen activity) Alkalinity m Total inorganic carbon (TIC) m HCO <sub>2</sub> - m		1,01 0,000
pH change by electron balancing (Phreeq Carbon equilibrium pH (Hydrogen activity) Alkalinity m Total inorganic carbon (TIC) m CO <sub>2</sub> m HCO <sub>3</sub> - m	c)	0,000
Carbon equilibrium pH (Hydrogen activity) Alkalinity m Total inorganic carbon (TIC) m CO <sub>2</sub> m HCO <sub>3</sub> - m	c)	
pH (Hydrogen activity) Alkalinity m Total inorganic carbon (TIC) m CO <sub>2</sub> m HCO <sub>3</sub> - m		
Alkalinity m Total inorganic carbon (TIC) m CO₂ m HCO₂ - m		
Total inorganic carbon (TIC) m CO2 m HCO3 - m	_	7,91
CO2 m HCO3- m	1eq/kgw	3,26
HCO3 - m	1mol/kgw	3,34
	imol/kgw	0,10
	imol/kgw	3,19
	nmol/kgw	0,01
dpH by 0.1 mmol HCl / kgw		-0,28
Buffer capacity m	1mol/kgw /pH	0,28
Calcite equilibrium		
SI (calcite)		0,18
Equilibrium-pH (pHs or pH-Langelier)		7,73
	nmol/kgw	0,05
Calcite Precipitation Potential at 60 C m		0,20
Calcite Precipitation Potential at 100 C m	imol/kgw	0,46
E-mail address (optional):		
		d / Mall Excel

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### Aquatic Chemistry for Engineers

# Modeling water treatment trains and processes with PHREEQC

2 April 2015

Peter de Moel – Omnisys/TU Delft



